Studies on the Total Synthesis of Crotocembreneic Acid and Neocrotocembreneic Acid:A Perchloric Acid-induced Lactonization Reaction of Epoxy Ester

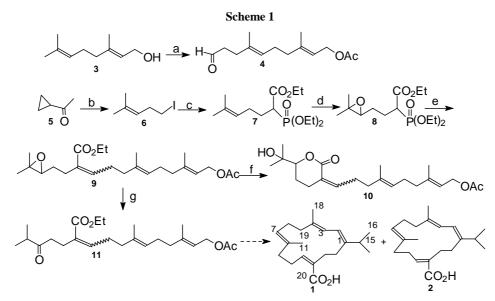
Zuo Sheng LIU, Jiong LAN, Li Zeng PENG, Yu Lin LI*

National Laboratory of Applied Organic Chemistry and Institute of Organic Chemistry, Lanzhou University, Lanzhou 730000

Abstract A perchoric acid-induced lactonization reaction of epoxy ester is described. Using lithium perchlorate, the ketone is obtained in 81% yield.

Keywords Lactonization, crotocembreneic acid, neocrotocembreneic acid, synthesis.

Cembranoids, a large family of diterpenoid natural products characterized by the presence of fourteen-membered ring, have been isolated from various marine sources as well as some terrestrial organisms since the 1960's¹. Crotocembraneic acid **1** and neocrotocembraneic acid **2** have been isolated from the stem bark of *Croton oblongifolius* Roxb. (Euphorbiaceae)². In our studies on the total synthesis of **1** and **2**, we find a novel perchloric acid-induced lactonization reaction of epoxy ester **9** (Scheme 1).



Reagents and conditions: a) ref. 3; b) i)MeMgI, Et₂O, reflux; ii)30%H₂SO₄, 0 °C, 1

Zuo Sheng LIU et al.

h, 77%; c) NaH, (Et₂O)P(O)CH₂CO₂Et, DMF, 60 °C, 3h, 88%; d) *m*-CPBA, CH₂Cl₂, 0 °C, 2 h, 93%; e) LDA, -78 °C, 1h, then aldehyde **4**, -78 °C~rt, 3h, 73%; f) HClO₄, CH₂Cl₂, 0 °C, 30min., 95%; g)LiClO₄, PhH, 80 °C, 1h, 81%.

Aldehyde **4** was prepared from geraniol **3** according to the reference 3. Using *Juvia* method, the homoprenyl iodide **6** was obtained from methyl cyclopropanyl ketone **5** in 77% yield⁴. The homoprenyl iodide reaction with sodium triethyl phosphonate gave phosphonate **7**. Epoxidation of phosphonate **7** with *m*-CPBA gave the epoxide **8** in 93% yield.

The *Horner-Emmons* reaction of phosphono ester **8** with aldehyde **4** using LDA as base at -78 °C gave α , β -unstatured ester **9** in 71% yield⁵. The ratio of *E* and *Z* isomers were *ca.* 1:1. As usual⁶, we used the perchloric acid to rearrange the epoxide to ketone⁷. To our surprise, the ketone **11** was obtained only in 5% yield, the lactone **10** was obtained in 95% yield, which the *E*, *Z* isomers were separated by silica gel chromatograph⁸. The rearrangement of epoxide **9** to ketone **10** reacted smoothly using lithium perchlorate⁹, the ketone **10** was obtained in 81% yield. The transformation of **10** to **1** and **2** is in progress.

Acknowledgment

This work was financially supported by the National Natural Science Foundation of China (Grant NO. 29672015)

References and Notes

- 1 M. A. Tius, Chem Rev. 1988, 88, 719.
- 2 S. Roengsumran, S. Archayindee, A, Petesom, K. Pudhom, P. Singtothong, C. Surachethapan, T. Vilaivan, *J. Nat. Prod.*, **1998**, *61*, 652.
- 3 Z. Liu, J. Lan, Z. Peng, Y. Li, *Chin. Chem. Lett.* **1999**, *10*, 547.
- 4 W. Biermacki, G. Amdrei, *Synthesis*, **1979**, 37.
- 5 G. B. Hammond, M. B. Cox, D. F. Wiemer, J. Org. Chem., 1990, 55, 128.
- 6 For review see: J. G. Smith, *Synthesis*, **1984**, 629.
- 7 J. Mao, Y. Li, Z. Hou, Y. Li, X. Liang, Scienca in China (Series B), 1992, 35, 257.
- 8 Spectral data: Compound **10E**: IR(film): v=3458, 2974, 2935, 1730, 1719, 1635, 1448, 1377, 1268, 1235, 1187, 960, cm⁻¹. EIMS (*m*/z): 378 (0.2%, M), 335 (0.2, M-Ac), 318 (1.1), 300 (0.4), 285 (0.3), 250 (0.6), 232 (1.7), 215 (1.3), 184 (10), 166 (16), 138 (16), 135 (14), 96 (16), 93 (22), 67 (25), 59 (58), 43 (100). ¹HNMR (400 MHz,CDCl₃): δ (ppm) 7.03 (t, J=6.7Hz, 1H, CH=), 5.35(t, J=7.2Hz, 1H, CH=), 5.13 (t, J=6.5Hz, 1H, CH=), 4.58 (d, J=7.2Hz, 2H, CH₂O), 4.03 (dd, J=11.6Hz, J=2.7Hz, 1H, CHO), 2.06 (s, 3H, CH₃), 2.36-2.00 (m, 12H, 6CH₂), 1.70 (s, 3H, CH₃), 1.61 (s, 3H, CH₃), 1.29 (s, 3H, CH₃), 1.26 (s, 3H, CH₃). ¹³CNMR (100 MHz, CDCl₃): δ (ppm) 171.09, 166.51, 146.16, 145.85, 125.76, 124.81, 119.23, 118.41, 85.51, 71.33, 61.33, 39.32, 37.67, 27.06, 26.07, 25.49, 24.40, 22.92, 22.61, 21.02, 16.43, 15.95. Compound **10Z**: ¹HNMR (400 MHz,CDCl₃): δ (ppm) 6.04 (t, J=6.7Hz, 1H, CH=), 5.33(t, J=7.2Hz, 1H, CH=), 5.12 (t, J=6.5Hz, 1H, CH=), 4.57 (d, J=7.2Hz, 2H, CH₂O), 4.02 (dd, J=11.7Hz, J=2.7Hz, 1H, CHO), 2.69 (m, 2H, CH₂), 2.04 (s, 3H, CH₃), 1.22 (s, 3H, CH₃). ¹³CNMR (100 MHz, CDCl₃): δ (ppm) 171.07, 165.41, 147.67, 147.24, 125.19, 124.44, 119.13, 118.30, 85.66, 71.32, 61.31, 39.42, 38.75, 28.64, 27.94, 26.11, 25.57, 24.28, 23.44, 20.99, 16.41, 15.77.
- 9 J. E. McMurry, J. H.Musser, M. S. Ahmad, L. C. Blaszczak, J. Org. Chem. 1975, 40, 1829.

Received 14 April 2000